

Table 3

The impact of dimethylenedioxide group on the thermolysis of peroxides

Peroxide	Solvent	Temperature, °C	Original concentration, mole/l	$k \cdot 10^{-4}$, c ⁻¹
I	benzene	50	$4 \cdot 10^{-2}$	0.35 ± 0.04
		60		1.20 ± 0.02
		70		3.30 ± 0.02
	ethylbenzene	50	$4 \cdot 10^{-2}$	0.69 ± 0.04
		60		1.96 ± 0.04
		70		4.38 ± 0.02
II	benzene	50	$4 \cdot 10^{-2}$	0.25 ± 0.06
		60		1.00 ± 0.05
		70		3.10 ± 0.02
	ethylbenzene	50	$4 \cdot 10^{-2}$	0.68 ± 0.03
		60		1.90 ± 0.06
		70		4.36 ± 0.04
III	benzene	50	$5 \cdot 10^{-2}$	0.50 ± 0.04
		60		1.70 ± 0.03
		70		5.32 ± 0.03
	ethylbenzene	50	$5 \cdot 10^{-2}$	0.39 ± 0.05
		60		1.08 ± 0.02
		70		3.30 ± 0.02
IV	benzene	50	$5 \cdot 10^{-2}$	0.40 ± 0.03
		60		1.42 ± 0.03
		70		5.20 ± 0.02
	ethylbenzene	50	$5 \cdot 10^{-2}$	0.36 ± 0.04
		60		1.08 ± 0.02
		70		2.98 ± 0.02